

Course Syllabus

1	Course title	Solid State Physics			
2	Course number	0302471			
3	Credit hours	3			
	Contact hours (theory, practical)	3			
4	Prerequisites/corequisites	Quantum Mechanics (0332361)			
5	Program title	Physics			
6	Program code	2			
7	Awarding institution	The University of Jordan			
8	School	Science			
9	Department	Physics			
10	Course level	Senior (fourth year)			
11	Year of study and semester(s)	First semester, 2023-2024			
12	Other department(s) involved in teaching the course	None			
13	Main teaching language	English			
14	Delivery method	\square Face to face learning \square Blended \square Fully online			
15	Online platforms(s)	\Box Moodle \boxtimes Microsoft Teams \Box Skype \Box Zoom			
		□Others			
16	Issuing/Revision Date	9/10/2023			



17 Course Coordinator:

Name: Riad Shaltaf	Contact hours: 11:30 – 13:00 Mon, Wed. Office number: 115
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18 Other instructors:

None

19 Course Description:

Crystal lattice and structure; reciprocal lattice; crystal binding; lattice vibrations; elastic scattering of waves; thermal properties of solids; free-electron gas, energy bands in solids.



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Course aims and outcomes:

Physics of Materials-0332371

B- Students Learning Outcomes (SLOs):

For purposes of mapping the course SLOs to the physics program SLOs, at the successful completion of the physics program, graduates are expected to be able to:

SLO (1) Master professionally a broad set of knowledge concerning the fundamentals in the basic areas of physics: Classical Mechanics, Electrostatics and Magnetism, Quantum Mechanics, Thermal Physics, Optics, Theory of Special Relativity, Mathematical Physics, Electronics.

SLO (2) Apply knowledge of mathematics and fundamental concepts in the basic areas of physics to identify and solve physics related problems.

SLO (3) Utilize computers and available software in both data collections and data analysis.

SLO (4) Utilize standard laboratory equipment, modern instrumentation, and classical techniques to design and conduct experiments as well as to analyze and interpret data.

SLO (5) Develop a recognition of the need and ability to engage in life-long learning.

SLO (6) Demonstrate ability to use techniques, skills, and modern scientific tools necessary for professional practice.

SLO (7) Communicate clearly and effectively in both written and oral forms.

SLO (8) Apply proficiently team-work skills and employ team-based learning strategies.

SLO (9) Apply professional and ethical responsibility to society.

Upon successful completion of this course, students will be able to:

Program SLOs	SLO								
Course SLOs	(1)	(2)	(3)	(4)	(5)	(6)	(7)	(8)	(9)
 Grasping the fundamentals of atomic arrangements, including crystal structures and defects, to comprehend material properties at the atomic level. 	~	~							
2. Grasp of fundamental material properties like mechanical, electrical, thermal, and optical properties, and the factors influencing these properties.	~	~							
3. Developing knowledge about how materials deform, fracture, and behave under stress, covering	~	~							

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	concepts like elasticity, plasticity, and toughness.						
4.	Understanding heat transfer mechanisms in materials and their response to changes in temperature, including thermal conductivity and expansion.	~	~				
5.	Familiarization with various experimental methods used to analyze and characterize materials, such as spectroscopy, microscopy, and diffraction.		~				
6.	Development of problem- solving skills through the application of principles learned to analyze and solve material-related issues or scenarios.	✓	~				

Week	Lecture	Торіс	Intended Learning Outcome	Learning Methods (Face to Face/Blended/ Fully Online)	Platform	Synchronous / Asynchrono us Lecturing	Evaluation Methods	Resources
1 9/10	1.1	Crystal structure	SLO(1),	Face to face			Test	For all units of this course, the students may
	1.2	Crystal structure	SLO(1),	Face to face			Test	refer to the following
2 16/10	2.1	Crystal structure	SLO(1),	Face to face			Test	resources:
	2.2	Crystal Structure	SLO(1),	Face to face			Test	*Text book and suggested references
3 23/10	3.1	Crystal Structure	SLO(1),	Face to face			Test	*Classroom
	3.2	Reciprocal Lattice	SLO(1), SLO(5)	Face to face			Test	
4 30/10	4.1	Reciprocal lattice	SLO(1), SLO(5)	Face to face			Test	



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	4.2	Reciprocal lattice	SLO(1), SLO(5)	Face to face		Test	
5 6/11	5.1	Reciprocal Lattice	SLO(1), SLO(5)	Face to face		Test	
	5.2	Reciprocal lattice	SLO(1), SLO(5)	Face to face		Test	
6 13/11	6.1	Crystal binding	SLO(3),	Face to face		Test	
	6.2	Crystal binding	SLO(3),	Face to face		Test	
7 20/11	7.1	Crystal binding	SLO(3),	Face to face		Test	
20/11	7.2	Crystal	SLO(3),	Face to face		Test	
		binding					
8	8.1	28/11/2022	, First Exam				
27/11							
27/11	8.2	Phonon modes	SLO(2), SLO(3)	Face to face		Test	
9 4/12	9.1	Phonon modes	SLO(2), SLO(3)	Face to face		Test	
	9.2	Phonon modes	SLO(2), SLO(3)	Face to face		Test	
10 11/12	10.1	Thermal properties	SLO(4),	Face to face		Test	
	10.2	Thermal properties	SLO(4),	Face to face		Test	
11 18/12	11.1	Thermal properties	SLO(4),	Face to face		Test	
	11.2	Thermal properties	SLO(4),	Face to face		Test	
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12	12.1	26/12/2022, 5	Second Exa	n				
25/12	12.2	Free electron gas	SLO(2),	Face to face			Test	
13 1/1	13.1	Free electron gas	SLO(2),	Face to face			Test	
	13.2	Free electron gas	SLO(2),	Face to face			Test	
14 8/1	14.1	Free electron gas	SLO(2),	Face to face			Test	
	14.2	Energy bands	SLO(2),	Face to face			Test	
15 15/1	15.1	Energy bands	SLO(2),	Face to face			Test	
	Tuesday	, 17/01/2023 is	the last day	of teaching. 19	/01/2023 -	- 30/01/2023	is the period	of final exa

22 Evaluation Methods:

Opportunities to demonstrate achievement of the SLOs are provided through the following assessment methods and requirements:

Evaluation Activity	Mark	Topic(s)	SLOs	Period (Week)	Platform
First Exam	20	Crystal structure & Reciprocal lattice	SLO(1), SLO(5)	Week 8 (28/11/2022)	
Second Exam	30	Crystal binding & Phonon modes	SLO(2) – SLO(4)	Week 12 (26/12/2022)	



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Final Exam	50	All course content	SLO(1) – SLO(5)		
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23 Course Requirements

Each student should have access to a computer & internet connection

24 Course Policies:

A- Attendance policies:

Attendance is mandatory. Students who record absences more than the legally acceptable limit **may lose their chance to sit for the final exam of the course**.

B- Absences from exams and submitting assignments on time:

False medical reports and other devious ways to avoid taking exams on time are not acceptable. The students are encouraged to handle their responsibility and develop **positive learning attitudes**.

C- Health and safety procedures:

Follow the instructions regarding health and safety procedures in the university.

D- Honesty policy regarding cheating, plagiarism, misbehavior:

The course is designed to provide students with learning opportunities. Group work and discussions accompanied with individual input and hard work are encouraged to fulfill the objectives of the course, whereas **cheating and misbehavior are completely unacceptable**.

E- Grading policy:

Do not waste time arguing about grades and grading policies. Instead, invest your time in fruitful learning.

F- Available university services that support achievement in the course:

- □ E-learning resources
- I Microsoft Teams
- I Smart Class rooms
- Computer facilities

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25 References:

A- Required book(s), assigned reading and audio-visuals:

*Textbook: Charles Kittel; Introduction to Solid State Physics, 8th Ed., (John Wiley & Sons, Hoboken, NJ, 2005) Power point presentations and illustrations designed to explain the contents of the course.

B- Recommended books, materials, and media:

[1] J.S. Blakemore, Solid State Physics, 2nd Ed., (Cambridge University Press, Cambridge, 1985)

[2] M.S. Rogalski and S.B. Palmer, Solid State Physics, (Gordon and Breach Science Publishers, Australia, 2000)

[3] J. Richard Christman, Fundamentals of Solid State Physics, John Wiley & Sons, New York, 1988)

26 Additional information:

Name of Course Coordinator: Riad Shaltaf	Signature	Date: 9/10/2024
Head of Curriculum Committee/Department:		Signature:
Head of Department:		- Signature:
Head of Curriculum Committee/Faculty:		Signature:
Dean:	Signature:	